

# Nitrate assimilation is inhibited by elevated CO<sub>2</sub> in field-grown wheat

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**Total protein and nitrogen concentrations in plants generally decline under elevated CO<sub>2</sub> atmospheres<sup>1,2</sup>. Explanations for this decline include that plants under elevated CO<sub>2</sub> grow larger, diluting the protein within their tissues<sup>3,4</sup>; that carbohydrates accumulate within leaves, downregulating the amount of the most prevalent protein Rubisco<sup>2</sup>; that carbon enrichment of the rhizosphere leads to progressively greater limitations of the nitrogen available to plants<sup>4</sup>; and that elevated CO<sub>2</sub> directly inhibits plant nitrogen metabolism, especially the assimilation of nitrate into proteins in leaves of C<sub>3</sub> plants<sup>5</sup>. Recently, several meta-analyses have indicated that CO<sub>2</sub> inhibition of nitrate assimilation is the explanation most consistent with observations<sup>6–8</sup>. Here, we present the first direct field test of this explanation. We analysed wheat (*Triticum aestivum* L.) grown under elevated and ambient CO<sub>2</sub> concentrations in the free-air CO<sub>2</sub> enrichment experiment at Maricopa, Arizona. In leaf tissue, the ratio of nitrate to total nitrogen concentration and the stable isotope ratios of organic nitrogen and free nitrate showed that nitrate assimilation was slower under elevated than ambient CO<sub>2</sub>. These findings imply that food quality will suffer under the CO<sub>2</sub> levels anticipated during this century unless more sophisticated approaches to nitrogen fertilization are employed.**

Many lines of evidence from laboratory studies demonstrate that elevated CO<sub>2</sub> concentrations in the atmosphere inhibit leaf nitrate (NO<sub>3</sub><sup>-</sup>) assimilation in C<sub>3</sub> plants. These include: plants receiving NO<sub>3</sub><sup>-</sup> as their sole source of nitrogen (N) accumulate less organic N under elevated than ambient CO<sub>2</sub> (refs 7,9–11); plants subjected to a pulse of <sup>15</sup>N–NO<sub>3</sub><sup>-</sup> incorporate less <sup>15</sup>N into organic N compounds under elevated than ambient CO<sub>2</sub> (ref. 10); plant growth is slower under elevated than ambient CO<sub>2</sub> when NO<sub>3</sub><sup>-</sup> serves as the sole N source and faster when NH<sub>4</sub><sup>+</sup> serves as the sole N source<sup>5,12</sup>; ΔAQ (changes in the ratio of net CO<sub>2</sub> consumption to net O<sub>2</sub> evolution after shifting N nutrition from NH<sub>4</sub><sup>+</sup> to NO<sub>3</sub><sup>-</sup>), a real-time measure of leaf NO<sub>3</sub><sup>-</sup> assimilation, decreases with increasing leaf internal CO<sub>2</sub> concentration<sup>9,12</sup>; and maximum NO<sub>3</sub><sup>-</sup> reductase activity *in vitro* is usually less under elevated than ambient CO<sub>2</sub> (refs. 11–13). Verification of CO<sub>2</sub> inhibition of NO<sub>3</sub><sup>-</sup> assimilation in the field, however, is still lacking.

Here, we conducted chemical analyses of wheat (*Triticum aestivum* L.) grown in 1996 and 1997 under elevated or ambient atmospheric CO<sub>2</sub> concentrations in the free-air CO<sub>2</sub> enrichment (FACE) experiment at Maricopa, Arizona. This experiment originally assessed grain yield<sup>14</sup>, total N of green leaves<sup>15</sup>, grain protein<sup>16</sup> and soil N dynamics<sup>17</sup>. Leaf material collected from this experiment was stored on ice, transported to the laboratory, oven dried at 70 °C, stored in evacuated plastic bags that were

sealed in paint cans, and kept in a storeroom at the US Water Conservation Laboratory in Phoenix, Arizona, USA until air freighted to UC Davis for the chemical analyses described below. This preparation and storage of samples minimized changes over time in total N, nitrate and nitrogen isotope ratios<sup>18</sup>. What prompted these additional analyses of the Maricopa leaf material was the development of a new technique to assess the N isotope signature of NO<sub>3</sub><sup>-</sup> (ref. 19) as well as a new perspective about the interactions between elevated CO<sub>2</sub> and NO<sub>3</sub><sup>-</sup> assimilation<sup>5,9,10,12</sup>.

The values for leaf total N (Fig. 1) did not differ from those reported over a decade earlier<sup>15</sup>, supporting the assumption that the samples were well preserved. Plants in the low-N treatment of either CO<sub>2</sub> treatment contained no detectable leaf NO<sub>3</sub><sup>-</sup> on most sampling dates (data not shown). In contrast, a significant percentage of leaf N remained as unassimilated NO<sub>3</sub><sup>-</sup> (ratio of NO<sub>3</sub><sup>-</sup> to total N) in plants subjected to the high-N treatment (Fig. 2). The first fertilization, applied 4 weeks after plant emergence, increased leaf NO<sub>3</sub><sup>-</sup> concentration in the short term an average of fivefold and twofold in 1996 and 1997, respectively (Figs 1 and 2). Therefore, we focused our analysis on the high-N treatment from week 6 onwards.

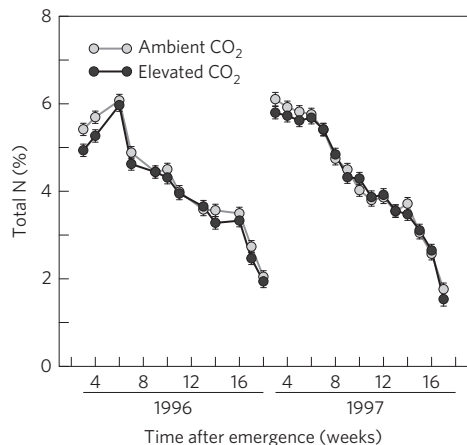
Leaf total N in the high-N treatment did not differ significantly between the CO<sub>2</sub> treatments ( $P = 0.12$ ; Fig. 1), but overall the ratio of NO<sub>3</sub><sup>-</sup> to total N was greater under elevated than ambient CO<sub>2</sub> from week 6 onwards ( $P < 0.0001$ ; Fig. 2). Total N and the ratio of NO<sub>3</sub><sup>-</sup> to total N were lower in 1996 than in 1997 ( $P < 0.003$ ; Figs 1 and 2). The analysis of variance tables are available in Supplementary Tables 2–5.

In leaves, both organic N and unassimilated NO<sub>3</sub><sup>-</sup> in 1996 were less enriched in <sup>15</sup>N under elevated than ambient CO<sub>2</sub> from week 6 onwards ( $P < 0.0001$ ; Figs 3 and 4). The δ<sup>15</sup>N of leaf organic N and NO<sub>3</sub><sup>-</sup> declined as the plants matured under both CO<sub>2</sub> treatments ( $P < 0.0001$ ; Figs 3 and 4).

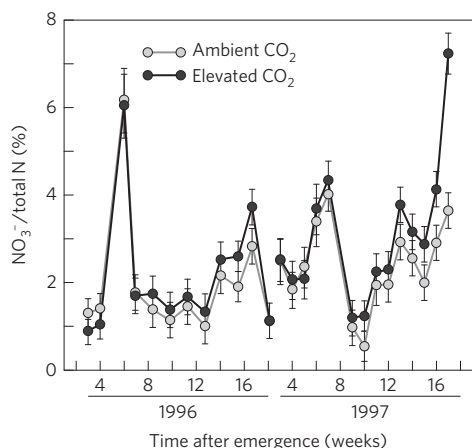
All three measures of NO<sub>3</sub><sup>-</sup> assimilation assessed in this study confirm that elevated CO<sub>2</sub> inhibited leaf NO<sub>3</sub><sup>-</sup> assimilation in field-grown wheat. The first measure was the proportion of leaf N that remained as free NO<sub>3</sub><sup>-</sup>. Leaf total N in the high-N treatment did not differ significantly between the CO<sub>2</sub> treatments (Fig. 1), as reported earlier<sup>15</sup>. The percentage of leaf total N that remained as unassimilated NO<sub>3</sub><sup>-</sup> was higher under elevated than ambient CO<sub>2</sub> from week 6 onwards in both years (Fig. 2 and Supplementary Tables 3–5). Higher free NO<sub>3</sub><sup>-</sup> relative to total N suggests that NO<sub>3</sub><sup>-</sup> assimilation was slower under elevated CO<sub>2</sub>.

The second measure was the δ<sup>15</sup>N of leaf organic N. It was more depleted in <sup>15</sup>N under elevated than ambient CO<sub>2</sub> from 6 weeks onwards (Fig. 3). If NO<sub>3</sub><sup>-</sup> availability does not limit assimilation, leaves preferentially assimilate <sup>14</sup>N–NO<sub>3</sub><sup>-</sup> (ref. 20). Therefore, the lower leaf δ<sup>15</sup>N<sub>organic</sub> signatures under elevated than ambient CO<sub>2</sub>

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**Figure 1 | Total nitrogen (percentage of dry matter) in wheat leaves as a function of time after emergence (weeks).** Shown are data from the 1996 and 1997 field seasons for plants grown under ambient ( $363 \mu\text{mol mol}^{-1}$  in 1996 and  $370 \mu\text{mol mol}^{-1}$  in 1997) or elevated ( $548 \mu\text{mol mol}^{-1}$  in 1996 and  $559 \mu\text{mol mol}^{-1}$  in 1997)  $\text{CO}_2$  atmospheres in the high-N fertilization treatment at Maricopa, Arizona. Plotted are the estimated means and standard errors (SAS Proc Mixed version 9.3).



**Figure 2 | Nitrate as a percentage of total N in wheat leaves as a function of time after emergence (weeks).** Shown are data from the 1996 and 1997 field seasons for plants grown under ambient ( $363 \mu\text{mol mol}^{-1}$  in 1996 and  $370 \mu\text{mol mol}^{-1}$  in 1997) or elevated ( $548 \mu\text{mol mol}^{-1}$  in 1996 and  $559 \mu\text{mol mol}^{-1}$  in 1997)  $\text{CO}_2$  atmospheric conditions in the high-N fertilization treatment at Maricopa, Arizona. Plotted are the estimated means and standard errors (SAS Proc Mixed version 9.3).

(Fig. 3) indicate that leaf  $\text{NO}_3^-$  assimilation was slower relative to replenishment of leaf  $\text{NO}_3^-$  from roots under elevated than ambient  $\text{CO}_2$ .

The third measure was  $\delta^{15}\text{N}$  of free  $\text{NO}_3^-$  in the leaves. If leaf  $\text{NO}_3^-$  assimilation is slower relative to replenishment of leaf  $\text{NO}_3^-$  from roots under elevated than ambient  $\text{CO}_2$ ,  $\text{NO}_3^-$  assimilation more slowly depletes leaf tissues of  $^{14}\text{N}-\text{NO}_3^-$  and  $\delta^{15}\text{N}_{\text{nitrate}}$  becomes less enriched in  $^{15}\text{N}$ . Here, unassimilated  $\text{NO}_3^-$  in wheat leaves was less enriched in  $^{15}\text{N}$  under elevated than ambient  $\text{CO}_2$  from 6 weeks onwards (Fig. 4), indicating that leaf  $\text{NO}_3^-$  assimilation was slower under elevated than ambient  $\text{CO}_2$ .

The isotopic signature of free  $\text{NO}_3^-$  in leaves also depends on the  $\delta^{15}\text{N}$  of  $\text{NO}_3^-$  translocated from the roots. For example, if  $\text{NO}_3^-$  assimilation rates in the roots are faster under elevated than ambient  $\text{CO}_2$  (ref. 21), isotope discrimination by nitrate reductase will enrich the root  $\text{NO}_3^-$  pool in  $^{15}\text{N}$ , and so  $\text{NO}_3^-$  translocated to the leaves

will be more  $^{15}\text{N}$  enriched. The  $\delta^{15}\text{N}$  of leaf  $\text{NO}_3^-$ , however, was lower under elevated than ambient  $\text{CO}_2$  (Fig. 4), indicating that the isotopic signature of  $\text{NO}_3^-$  derived primarily from leaf  $\text{NO}_3^-$  assimilation being slower under elevated than ambient  $\text{CO}_2$ .

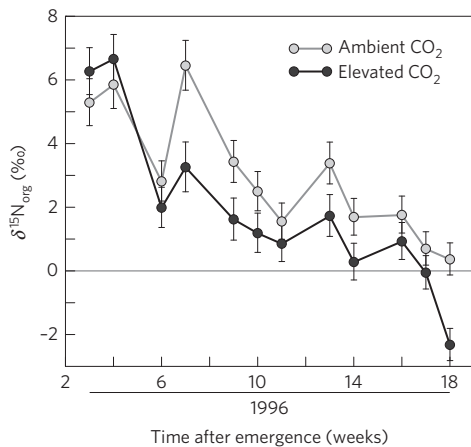
These field results are consistent with those of previous laboratory studies showing that several physiological mechanisms are responsible for  $\text{CO}_2$  inhibition of leaf  $\text{NO}_3^-$  assimilation in  $\text{C}_3$  plants<sup>5,9,10,12</sup>. One mechanism involves the first biochemical step of  $\text{NO}_3^-$  assimilation, the conversion of  $\text{NO}_3^-$  to  $\text{NO}_2^-$  in the cytoplasm of leaf mesophyll cells. Photorespiration stimulates the export of malic acid from chloroplasts<sup>22</sup> and increases the availability of NADH in the cytoplasm<sup>23</sup> that powers this first step<sup>24,25</sup>. Elevated  $\text{CO}_2$  decreases photorespiration and thereby decreases the amount of reductant available for  $\text{NO}_3^-$  reduction. Another physiological mechanism is that elevated  $\text{CO}_2$  inhibits  $\text{NO}_2^-$  influx into chloroplasts, and this decreases  $\text{NO}_3^-$  assimilation<sup>12</sup>. A third physiological mechanism is that processes in the chloroplast stroma compete for reduced ferredoxin: because ferredoxin-NADP reductase has a higher affinity for reduced ferredoxin than nitrite reductase<sup>26</sup>,  $\text{NO}_3^-$  assimilation proceeds only if the availability of reduced ferredoxin exceeds that needed for NADPH formation<sup>24,27</sup>. For most plants, this occurs when  $\text{CO}_2$  availability limits  $\text{C}_3$  carbon fixation<sup>10</sup>.

Several earlier studies at the Maricopa FACE site examined soil N in wheat plots that received irrigation, fertilizer and  $\text{CO}_2$  treatments similar to the high-N treatment here. Total inorganic N through the soil profile was similar in the ambient and elevated  $\text{CO}_2$  treatments from 6 weeks onwards<sup>28</sup>. Nitrogen mineralization was unaffected by  $\text{CO}_2$  treatment<sup>17</sup>, and soil  $\text{NO}_3^-$  constituted between 90 and 98% of inorganic N extractable by 2 M KCl at harvest (S. A. Prior and H. A. Torbert personal communication, 2013). Therefore, these soil N data support the conclusion that the leaf N differences that we observed between the elevated and ambient  $\text{CO}_2$  treatments derived from altered plant responses and not altered soil N availability.

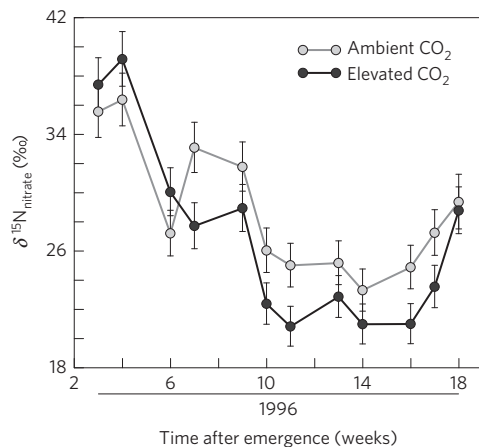
Several recent meta-analyses of the literature on plant responses to elevated  $\text{CO}_2$  support that  $\text{CO}_2$  inhibits leaf  $\text{NO}_3^-$  assimilation. One<sup>7</sup> based on 43 studies of wheat protein and grain yield under ambient and elevated  $\text{CO}_2$  concluded that 'elevated  $\text{CO}_2$  has a direct negative effect on grain protein accumulation independent of the yield effect, supporting recent evidence of  $\text{CO}_2$ -induced impairment of nitrate uptake/assimilation'. Another meta-analysis<sup>6</sup> based on 38 studies of soil  $\text{NH}_4^+$  and  $\text{NO}_3^-$  concentrations and plant  $\text{NH}_4^+$  and  $\text{NO}_3^-$  uptake in 58 species concluded that 'differential  $\text{CO}_2$  effects on soil  $\text{NH}_4^+$  and  $\text{NO}_3^-$ ... were consistent qualitatively with recent discoveries of  $\text{eCO}_2$  effects on plant N utilization', citing our laboratory studies on  $\text{CO}_2$  inhibition of leaf  $\text{NO}_3^-$  assimilation.

Under elevated  $\text{CO}_2$ , protein concentrations in wheat grain<sup>16,29</sup>, rice grain<sup>8</sup>, potato tuber<sup>8</sup> and barley grain<sup>29,30</sup> decline an average of around 8%. Wheat, rice, potato and barley, respectively, provide 21, 13, 2 and 0.3% of the protein in the human diet<sup>31</sup>. Consequently, protein available for human consumption may diminish by about 3% as atmospheric  $\text{CO}_2$  reaches the levels anticipated during the next few decades.

Increased yields under  $\text{CO}_2$  enrichment and heavy N fertilization may partially compensate for the decrease in food quality resulting from elevated  $\text{CO}_2$ . In the low-N treatment at Maricopa, elevated  $\text{CO}_2$  increased grain yields by 9% (ref. 32), but decreased grain protein concentrations by 11% (ref. 16), and so grain protein yields decreased by about 2%. In the high-N treatment, elevated  $\text{CO}_2$  increased grain yields about 16% (ref. 32), but had an insignificant effect on grain protein concentrations<sup>16</sup>, and so grain protein yields increased about 16%. In the high-N treatment, however, the fertilizer and soil supplied 430 and 490  $\text{kg N ha}^{-1}$  during 1996 and 1997, respectively, but crop biomass N under elevated  $\text{CO}_2$  was only



**Figure 3 | Isotopic signature of organic N ( $\delta^{15}\text{N}_{\text{org}}$ ) in wheat leaves as a function of time after emergence (weeks).** Shown are data from the 1996 field season for plants grown under ambient ( $363 \mu\text{mol mol}^{-1}$ ) and elevated ( $548 \mu\text{mol mol}^{-1}$ )  $\text{CO}_2$  atmospheres in the high-N fertilization treatment at Maricopa, Arizona. Plotted are the estimated means and standard errors (SAS Proc Mixed version 9.3).



**Figure 4 | Isotopic signature of nitrate ( $\delta^{15}\text{N}_{\text{nitrate}}$ ) in wheat leaves as a function of time after emergence (weeks).** Shown are data from the 1996 field season for plants grown under ambient ( $363 \mu\text{mol mol}^{-1}$ ) and elevated ( $548 \mu\text{mol mol}^{-1}$ )  $\text{CO}_2$  atmospheres in the high-N fertilization treatment at Maricopa, Arizona. Plotted are the estimated means and standard errors (SAS Proc Mixed version 9.3).

$270 \text{ kg N ha}^{-1}$ , indicating that nearly half of the N applied was not retained in the crop<sup>16,33</sup>. Thus, such high fertilization rates would be undesirable because of higher costs, greater  $\text{NO}_3^-$  leaching into groundwater, and greater  $\text{N}_2\text{O}$  emissions.

Obviously, plants have alternative strategies for acquiring organic N. One such strategy is the use  $\text{NH}_4^+$  as a N source. Undoubtedly, the wheat at Maricopa absorbed  $\text{NH}_4^+$  as part of its mineral N supply, but nitrification at this site was rapid<sup>17</sup>, and the soils contained relatively little  $\text{NH}_4^+$  as the growing season progressed (S. A. Prior and H. A. Torbert personal communication, 2013). Another strategy is root  $\text{NO}_3^-$  assimilation, which may be enhanced under elevated  $\text{CO}_2$ <sup>21</sup>. Unfortunately, relatively little is known about the extent to which the balance between root and leaf  $\text{NO}_3^-$  assimilation varies within and among species<sup>34</sup>. Breeding crops for enhanced root  $\text{NO}_3^-$  and  $\text{NH}_4^+$  assimilation has the potential to compensate for lower shoot  $\text{NO}_3^-$  assimilation rates and likely losses in food quality as atmospheric  $\text{CO}_2$  rises, but this approach is yet untested.

## Methods

Wheat (*Triticum aestivum* L.) leaves were obtained from the 1996 and 1997 FACE experiment at the Maricopa Agricultural Center near Phoenix, Arizona<sup>15</sup>. Briefly, high- and low-N treatments at this site were assigned in four replicates under ambient and FACE rings 25 m in diameter. Within the rings, ambient and elevated  $\text{CO}_2$  were controlled at  $363$  and  $548 \mu\text{mol CO}_2 \text{ mol}^{-1}$  in 1996 and  $370$  and  $559 \mu\text{mol CO}_2 \text{ mol}^{-1}$  in 1997 by releasing air containing different  $\text{CO}_2$  concentrations from 2.5-m-high vertical pipes spaced every 2 m around the periphery. Certified seed of Yecora Roho, a cultivar still widely used in the region<sup>35</sup>, was planted on 15 December 1995 or 1996 and seedlings emerged 1 January 1996 or 1997. The soil at the experimental site is classified as Trix clay loam, fine-loamy, mixed (calcareous), hyperthermic Typic Torrifuvents. Nitrogen fertilizer in the form of  $\text{NH}_4\text{NO}_3$  was applied in the drip irrigation water: the high-N treatment received four applications ( $50 \text{ kg N ha}^{-1}$  at 4 weeks after emergence,  $125 \text{ kg N ha}^{-1}$  at 8 weeks,  $125 \text{ kg N ha}^{-1}$  at 12 weeks, and  $50 \text{ kg N ha}^{-1}$  at 16 weeks) for a total rate of  $350 \text{ kg N ha}^{-1}$ ; the low-N treatment received a total of 70 and  $15 \text{ kg N ha}^{-1}$  for 1996 and 1997, respectively, in three increments<sup>16,35</sup>. In addition to the fertilizer applied, substantial residual inorganic N was present at sowing ( $80 \text{ kg N ha}^{-1}$  in 1996 and  $145 \text{ kg N ha}^{-1}$  in 1997).

Plant harvests were made at 10–14 day intervals through the season<sup>15</sup>. At each harvest, 24 plants were sampled within each replicate of a treatment. The plants were stored on ice and transported to the laboratory. Green leaf tissue was oven dried at  $70^\circ\text{C}$  and stored in evacuated plastic bags that were sealed in paint cans. Subsequently, this leaf tissue was ball-milled at UC Davis, and total N and total N isotope ratios were determined using an elemental analyser interfaced to an isotope ratio mass spectrometer (Sercon) at the UC Davis Stable Isotope Facility. During analysis, samples were interspersed with two or more different  $\delta^{15}\text{N}$  standards.

Nitrate was extracted with 1 mM  $\text{CaSO}_4$  from subsamples of the pulverized leaves by using an orbital shaker, followed by centrifugation. The  $\text{NO}_3^-$  concentration of the diluted extracts was determined spectrophotometrically<sup>36</sup>. The nitrogen isotopic composition of plant  $\text{NO}_3^-$  extracts was analysed from  $\text{N}_2\text{O}$  generated by denitrifying bacteria lacking  $\text{N}_2\text{O}$  reductase<sup>19</sup>. Briefly, 2 ml aliquots of *Pseudomonas chlororaphis* culture were sealed into 20 ml headspace vials that were purged for 2 h with  $\text{N}_2$  gas to remove  $\text{N}_2\text{O}$  and  $\text{O}_2$ . Samples containing  $0.1 \mu\text{mol NO}_3^- \text{ N}$  of the plant tissue extracts or standards were injected through the septae of the vials. The  $\text{N}_2\text{O}$  was flushed from the vials with He, trapped cryogenically, and then released into the isotope ratio mass spectrometer (Sercon) at the UC Davis Stable Isotope Facility. Samples were interspersed with  $\delta^{15}\text{N}$   $\text{KNO}_3$  standards that were processed like the plant tissue extracts.

The leaf samples collected in the second year (1997) seemed to have become contaminated with the heavy nitrogen isotope because the  $\delta^{15}\text{N}$  values were highly variable and reached up to 250 ‰ in individual samples. Therefore, we report N isotope ratios for only the first year (1996).

Leaf organic N was estimated from the difference between leaf total N and leaf unassimilated  $\text{NO}_3^-$  because  $\text{NH}_4^+$  concentrations in wheat leaves are low and do not vary significantly with  $\text{CO}_2$  treatment<sup>12</sup>. The isotope ratio of the leaf organic nitrogen ( $\delta^{15}\text{N}_{\text{organic}}$ ) was thus calculated by dividing the mass of ( $^{15}\text{N}_{\text{total}} - ^{15}\text{N}_{\text{NO}_3^-}$ ) by the mass of ( $^{14+15}\text{N}_{\text{total}} - ^{14+15}\text{N}_{\text{NO}_3^-}$ ).

The first fertilization, applied 4 weeks after plant emergence, increased leaf  $\text{NO}_3^-$  concentration an average of fivefold and twofold in 1996 and 1997, respectively, and therefore, we considered only data collected after week 6 in our statistical analysis. An analysis of variance using the MIXED procedure with repeated measures in SAS (version 9.3, SAS Institute) assessed the effects of year (1 or 2 years),  $\text{CO}_2$  treatment (2 treatments), week after emergence (11 weeks), blocks (4 blocks), and their interactions on total N, the ratio of  $\text{NO}_3^-$  to total N,  $\delta^{15}\text{N}_{\text{nitrate}}$  and  $\delta^{15}\text{N}_{\text{organic}}$  (see Supplementary Tables for the SAS program used and the resulting analysis of variance tables). All of the data met the assumptions of normality and homogeneity of variance as evaluated using the Shapiro–Wilks and Levene's tests, respectively. The dependent variables (total N, ratio of  $\text{NO}_3^-$  to total N,  $\delta^{15}\text{N}_{\text{nitrate}}$  and  $\delta^{15}\text{N}_{\text{organic}}$ ) were repeated for each experimental block. The independent variables and their interactions were considered significant when  $P \leq 0.05$ .

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### Author contributions

All authors contributed to the data set, discussed the results and commented on the manuscript. A.J.B. and M.B. designed the study. M.B. conducted the chemical analyses. A.J.B. carried out the statistical analysis and wrote the paper.

### Additional information

Supplementary information is available in the online version of the paper. Reprints and permissions information is available online at [www.nature.com/reprints](http://www.nature.com/reprints). Correspondence and requests for materials should be addressed to A.J.B.

### Competing financial interests

The authors declare no competing financial interests.